

- [6] J. C. Sheehan, D. Ben-Ishai & J. U. Piper, J. Amer. chem. Soc. *95*, 3064 (1973).
 [7] R. Heymes, G. Amiard & G. Nominé, C. r. hebd. séances Acad. Sc. *263*, 170 (1966).
 [8] E. J. Corey, N. M. Weinshenker, T. K. Schaaf & W. Huber, J. Amer. chem. Soc. *91*, 5675 (1969).
 [9] R. Schwyzer, Helv. *35*, 1903 (1952).
 [10] K. D. Barrow & T. H. Spotwood, Tetrahedron Letters *1965*, 3325.
 [11] H. Schinz & M. Hinder, Helv. *30*, 1349 (1947).
 [12] J. W. Cornforth, (Mrs) R. H. Cornforth & K. K. Mathew, J. chem. Soc. *1959*, 112.
 [13] H. O. House & R. S. Ro, J. Amer. chem. Soc. *80*, 182 (1958).

244. Conversion of Penicillin to Cephalosporin Lactones

Preliminary Communication

by **Kapa K. Prasad**, **Gérard Schmid** and **Theodor Petrzilka**¹⁾

Organisch-chemisches Laboratorium der Eidgenössischen Technischen Hochschule Zürich,
 Universitätstrasse 16, CH-8006 Zürich

Dedicated to Professor R. B. Woodward on his 60th birthday

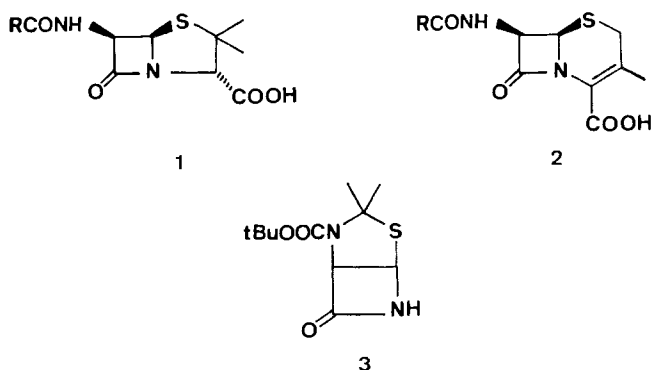
(18. VIII. 76)

Summary. Penicillin G (**1**, R = CH₂Ph) was converted to cephaloram- and 6-epicephaloram lactones **11** and **12** respectively by the initial replacement of thiazolidine part of penicillin by the mercaptan **9** followed by intramolecular cyclization and subsequent introduction of double bond at C(3).

Morin et al. [1] provided a useful method for the conversion of penicillin **1** into deacetoxy cephalosporin **2** through a novel process of oxidation and ring expansion.

In 1965 *Sheehan & Brandt* [2] have demonstrated the liberation of the β -lactam nitrogen from the thiazolidine part by *Curtius* degradation of the acid azide and hydrolysis. Furthermore they have achieved the complete replacement of thiazolidine

Scheme 1



part of penicillin molecule by nucleophiles such as mercaptans [3] [4], keeping intact the relatively labile β -lactam ring.

¹⁾ Author to whom correspondence should be addressed.

An interesting modification of the above degradation was introduced by *Woodward et al.* [5] for the conversion of penicillin to a fused thiazolidine **3** [6], which was one of the important intermediates in their classical total synthesis of cephalosporin [7].

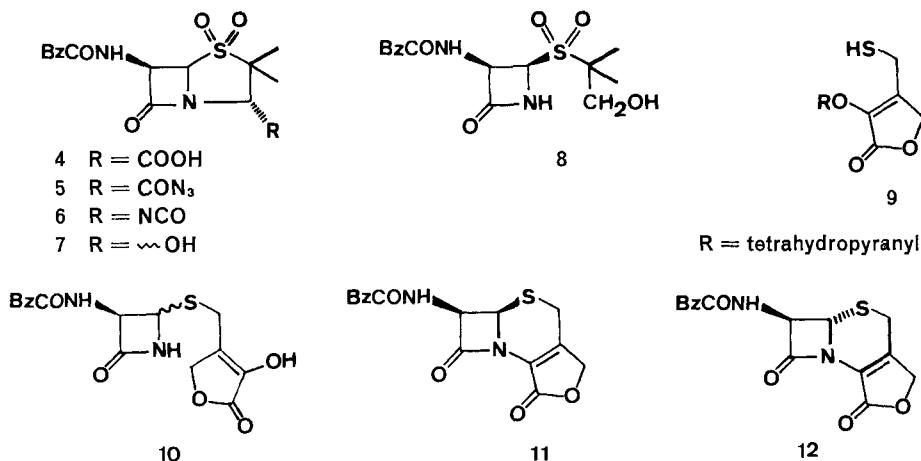
By replacing the thiazolidine part of the penicillin molecule by mercapto butenolide **9** and using our new method (see preceding communication) we converted penicillin G (**1**, R = CH₂Ph) into cephaloram- and 6-epicephaloram lactones **11** and **12** respectively.

Penicillin G (**1**, R = CH₂Ph) was oxidized to the sulfone **4** using potassium permanganate [8] [m.p. 124°, $[\alpha]_D^{25} = +167^\circ$ (acetone); IR. (KBr): 3345, 1805, 1745, 1635, 1550, 1520, 1320, 1115 cm⁻¹; NMR. (acetone-d₆): 8.30 (br., 1H, COOH), 7.49 (*d*, *J* = 10 Hz, 1H, NH), 7.28 (*m*, 5H, phenyl), 6.08 (*d* × *d*, *J*₁ = 4, *J*₂ = 10 Hz, 1H, H-C(6)), 5.13 (*d*, *J* = 4 Hz, 1H, H-C(5)), 4.44 (*s*, 1H, H-C(3)), 3.72 (*s*, 2H, benzyl CH₂), 1.57 (*s*, 3H, CH₃), 1.46 (*s*, 3H, CH₃)].

Sulfone **4** was converted to the azide **5** [IR. (CHCl₃): 3410, 2115, 1815, 1690, 1310 cm⁻¹] *via* the mixed anhydride using pyridine and isobutylchloroformate followed by sodium azide treatment. Thermal rearrangement of the azide **5** gave the isocyanate **6** [IR. (CHCl₃): 3405, 2260, 1815, 1690, 1310 cm⁻¹], which on acid hydrolysis gave 3-hydroxyphenam **7** [m.p. 111–112°, $[\alpha]_D^{25} = +77^\circ$ (acetone); IR. (CHCl₃): 3580, 3400, 1805, 1630, 1325 cm⁻¹; NMR. (CDCl₃): 7.22 (*m*, 5H, phenyl), 7.07 (*d*, *J* = 10 Hz, 1H, NH), 5.96 (*d* × *d*, *J*₁ = 4, *J*₂ = 10 Hz, 1H, H-C(6)), 5.16 (br., *s*, 2H, H-C(3) and OH, on D₂O exchange, *s*, 1H), 4.73 (*d*, *J* = 4 Hz, H-C(5)), 3.59 (*s*, 2H, benzyl CH₂), 1.38 (*s*, 6H, two CH₃)].

Compound **7** was reduced to azetidinone alcohol **8** with sodium borohydride [oil, $[\alpha]_D^{25} = +37^\circ$ (acetone); IR. (CHCl₃): 3400, 1780, 1670, 1300 cm⁻¹; NMR. (CDCl₃): 7.51 (*s*, 1H, H-C(1)), 7.26 (*s*, 5H, phenyl), 7.13 (*d*, *J* = 10 Hz, 1H, C(3) NH), 5.78 (*d* × *d*, *J*₁ = 4, *J*₂ = 10 Hz, H-C(3)), 5.06 (*d*, *J* = 4 Hz, 1H, H-C(4)), 3.58 (*s*, 2H, benzyl CH₂), 3.94–3.28 (br., 2H, on D₂O exchange collapsed into a clean *AB*-quartet,

Scheme 2



$J = 13$ Hz, 2H, CH₂), 2.10 (br., 1H, D₂O exchangeable, OH), 1.26 (s, 3H, CH₃), 1.23 (s, 3H, CH₃)].

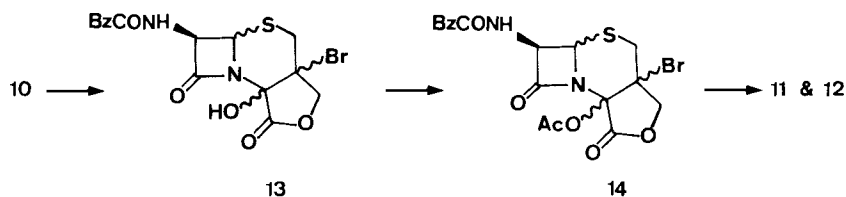
Reaction of compound **8** with mercaptan **9** in presence of triethylamine gave compound **10** as a mixture of isomers in 90% yield. This mixture on successive treatment with (i) bromine in presence of sodium hydrogencarbonate, (ii) acetyl chloride and pyridine, and (iii) zinc and acetic acid gave cephaloram lactone (**11**) and 6-epicephaloram lactone (**12**) in the ratio of 1:1 in 30% yield starting from the azetidinone **8**.

Cephaloram lactone 11: m.p. 210°, $[\alpha]_D^{25} = +142^\circ$ (acetone). – IR. (KBr): 3240, 1790, 1775, 1755, 1655 cm⁻¹. – NMR. (acetone-d₆): 8.06 (br., 1H, NH); 7.34 (s, 5H, phenyl); 5.97 (*d* × *d*, $J_{cis} = 5$, $J_{H, NH} = 9$ Hz, 1H, H-C(7)), 5.16 (*d*, $J_{cis} = 5$ Hz, 1H, H-C(6)); 5.04 (s, 2H, lactone CH₂); 3.83 (*q*, $J_{gem} = 18$ Hz, 2H, SCH₂); 3.68 (s, 2H, benzyl CH₂). – UV. $\lambda_{max}^{EtOH}(\epsilon)$: 259 (9940) nm. All these Data were found to be identical in all respects with the authentic sample obtained from the phenacetylation of 7-aminocephalosporin lactone.

6-Epicephaloram lactone 12: m.p. 176–177°, $[\alpha]_D^{25} = -95^\circ$ (acetone). – IR. (KBr): 3370, 1785, 1745, 1670 cm⁻¹. – NMR. (Acetone-d₆): 8.18 (br., 1H, NH); 7.33 (s, 5H, phenyl); 5.03 (s, 2H, lactone CH₂); 4.91 (*d* × *d*, $J_{trans} = 2.5$, $J_{H, NH} = 8$ Hz, 1H, H-C(7)); 4.83 (*d*, $J_{trans} = 2.5$ Hz, 1H, H-C(6)); 3.81 (*q*, $J_{gem} = 18$ Hz, 2H, SCH₂); 3.64 (s, 2H, benzyl CH₂). – UV. $\lambda_{max}^{EtOH}(\epsilon)$: 262 (9730) nm.

In analogy with our results on a related cyclization (see preceding communication) we assume the formation of intermediates **13** and **14** involving the mixture of stereoisomers in the transformation of **10** to compounds **11** and **12** (Scheme 3).

Scheme 3



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REFERENCES

- [1] *R. B. Morin, B. G. Jackson, R. A. Mueller, E. R. Lavagnino, W. B. Scanlon & S. L. Andrews*, J. Amer. chem. Soc. *91*, 1401 (1969).
- [2] *J. C. Sheehan & K. G. Brandt*, J. Amer. chem. Soc. *87*, 5468 (1965).
- [3] *J. C. Sheehan & C. A. Panetta*, J. org. Chemistry *38*, 940 (1973).
- [4] *J. C. Sheehan, H. C. Dalzell, J. H. Greenwood & D. R. Ponzi*, J. org. Chemistry *39*, 277 (1974).
- [5] *K. Heusler & R. B. Woodward*, German Offen. 1970, 935607.
- [6] *K. Heusler*, Helv. *55*, 388 (1972).
- [7] *R. B. Woodward, K. Heusler, J. Gosteli, P. Naegeli, W. Oppolzer, R. Ramage, S. Ranganathan & H. Vorbruggen*, J. Amer. chem. Soc. *88*, 852 (1966).
- [8] *D. A. Johnson, C. A. Panetta & D. E. Cooper*, J. org. Chemistry *28*, 1927 (1963).